

vapor  
 changing oxygen see  
 water volume heat energy atoms typical undergo  
 chemists identify fact  
 oxygen see  
 still later contains unknown least indication part anything  
 whether reassembled lot  
 observed temperature seen  
 matter chemically either occur  
 unburned  
 vapor  
 gas definitely hardness ripping use  
 frequently mechanical  
 Materials exactly types formation  
 study see without solid learn  
 involved material  
 chemists left paper mass  
 texture  
 whereas  
 learn  
 chemists  
 left  
 paper  
 mass  
 matter  
 new  
 original basic  
 state  
 containing  
 tend  
 molecules  
 methane makes freeze  
 carbon  
 make  
 appearance  
 molecule  
 good rely  
 notice way  
 changed  
 food  
 CO2  
 act  
 chemical  
 difference studying structure burned  
 always form words  
 related  
 observe  
 nature undergoes likely  
 observations partially held  
 look  
 contrast CH4 case  
 trying actually burning  
 Physical Figure  
 Believe hydrogen color  
 chemicals H2O  
 turned  
 involves bonded  
 taste  
 burn  
 might smell  
 substances feel melt  
 directions  
 Physical Figure  
 Hydrogen color  
 chemicals H2O  
 bonded  
 turned  
 melting distinction thermal harder  
 forever weight different  
 change  
 however similar  
 ripped