

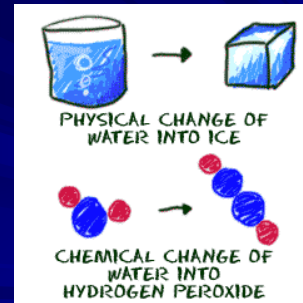
# TA01B - Teach About Matter and Changes

Use with BrishLab PS01B  
Done By: Coach

StarMaterials.com

1- What are the two types of properties that chemists use to study matter?

Page 1  
Para 3



Chemists look at Physical and Chemical Properties

Image Link

2- List five properties that chemists observe to identify matter.

Page 1  
Para 7



Smell, color, state, temperature, volume, mass and texture can be observed.

Image Link

3- What are the two major categories of matter changes?

Page 1  
Para 8



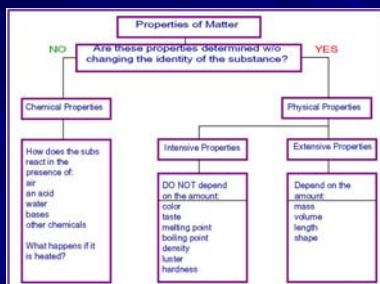
Physical and Chemical changes are ways to sort matter.

Image Link

Image Link

4- List seven observations of physical properties.

Page 2  
Para 10



Smell, color, state, temperature, volume, mass, texture, hardness and melting temperature are physical properties.

Image Link

5- Give an example of a physical change of water, H<sub>2</sub>O.

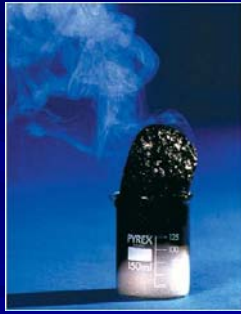
Page 2  
Para 11



Water boils in a Physical change.

Image Link

6- Why are **chemical changes** harder to reverse? Page 2  
Para 13



One thing changes into a **new** thing.

[Image Link](#)

7- What is a good **example** of a **chemical change**? Page 2  
Para 14



Burning paper makes ash in a **Chemical change**.

[Image Link](#)

8- What is the **major difference** between a **physical** and a **chemical change**? Page 2  
Para 15



In a **chemical change**, atoms form **new bonds** to make something **new**.

[Image Link](#)

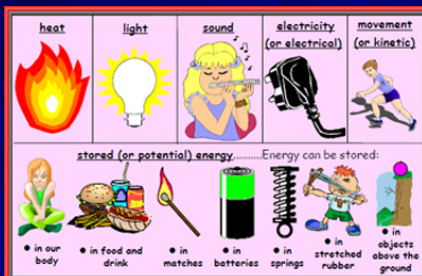
9- How many **times** can we change **ice** into **water** and back again into **ice**? Page 2  
Para 16



We can melt and freeze water forever - it is **reversible**.

[Image Link](#)

10- List **four kinds** of **energy** that can be used to **change matter**. Page \_\_\_  
Para \_\_\_



**Thermal changes, heat, electricity, chemical energy and mechanical energy** are always involved in a change.

[Image Link](#)

**Draw, label and color** a **physical change** and a **chemical change**.



[Image Link](#)

[Image Link](#)